

Intramolecular Baylis-Hillman Reaction

Dandamudi V Lenin

School of Chemical Sciences, Central University of Gujarat, Gandhinagar, Gujarat, India

Abstract

The Baylis–Hillman reaction is a novel atom economic, carbon-carbon bond forming three component reaction between activated alkene and carbon electrophile under catalytic influence of a catalyst/catalytic system. This review highlights developments of the intramolecular version of Baylis–Hillman adduct in the last three decades.

Keywords: Carbon-carbon bond forming reaction, intramolecular B-H reaction, biologically important molecules, 1,4-Diazabicyclo (2.2.2) octane

1. Introduction

The recent developments in organic chemistry mandate the nonstop convenient methodologies for carbon-carbon bonds formation and functional group transformation strategies with simple processes. The Baylis–Hillman reaction [1-3] is tertiary amine catalytic three component, atom economical carbon-carbon bond forming reaction. Baylis–Hillman reaction is a coupling reaction of α -position of activated alkenes with electrophiles under the influence of a tertiary amine catalyst, providing fascinating class of densely functionalized molecules (Fig. 1). The Baylis–Hillman reaction [4, 5] invention by American chemists A. B. Baylis and M. E. D. Hillman in the year 1973.

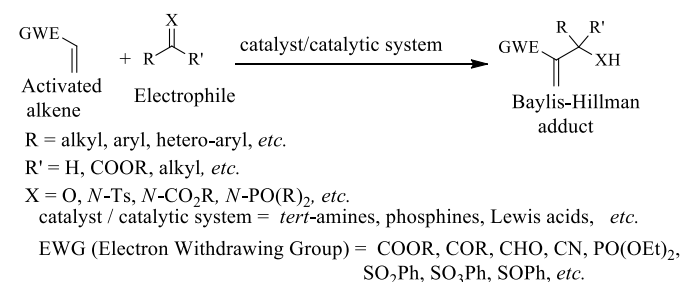
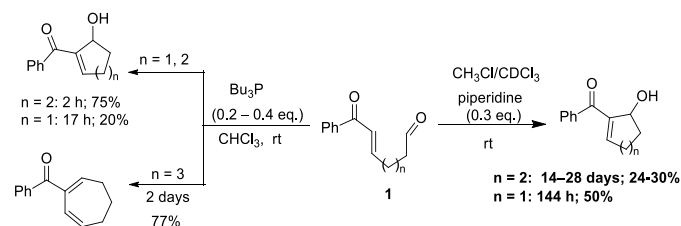


Fig. 1 General equation

Intramolecular Baylis-Hillman reaction

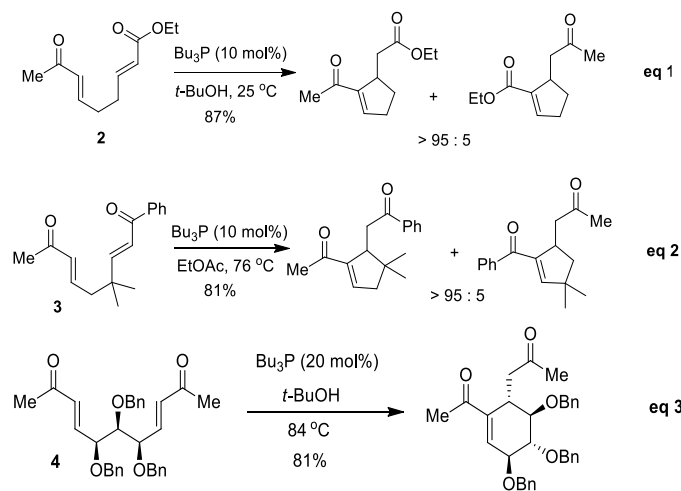
If the substrates contain both the activated alkene and electrophile components in appropriate position, there is possibility for performing intramolecular version of Baylis–Hillman reaction providing cyclic adducts. During the last several years this reaction has developed with respect to all the three essential compounds i.e. (1) Electrophiles, (2) Activated alkenes and (3) Catalysts or catalytic systems. Although there have been great advances in the case of all the three essential components, it is exciting to note that the intramolecular version remains at beginning. However, in this aspect has received more consideration from the synthetic chemists and significant progress has been reached. The interesting developments in this way are presented in this paper.

Murphy and co-workers [6, 8] have reported an intramolecular version of Baylis–Hillman reaction of enolaldehydes (1) in the presence of secondary amine (piperidine) or Bu₃P to offer the corresponding Baylis–Hillman adducts or eliminated cycloheptadiene (Scheme 1).

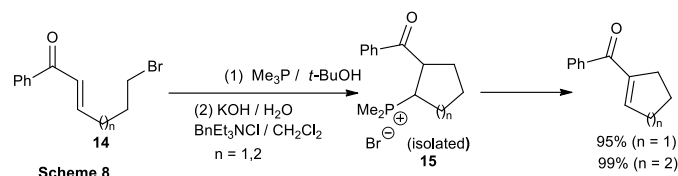


Scheme 1

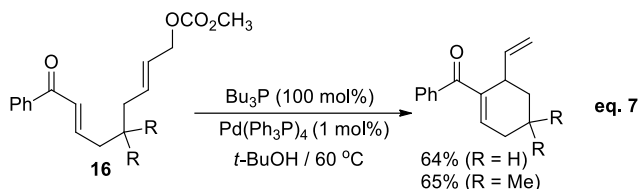
Krische and co-workers [9] published an interesting Bu₃P catalyzed cycloisomerization of bisenones (2 & 3). They have examined the effect of electronic (eq 1) and steric factors (eq 2) on cyclization. They have also further studied this methodology for the diastereoselective intramolecular Baylis–Hillman reaction of chiral bisenone (4) derived from xylose (eq 3) [9].



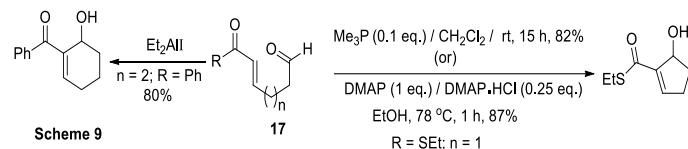
Roush and co-workers [10] reported an intramolecular



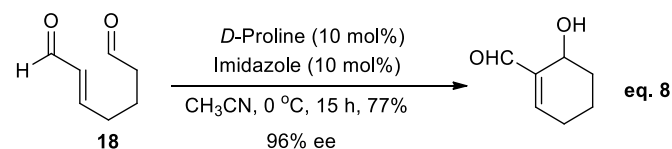
An intramolecular ring closing reaction of the enone-carbonates (16) under the influence of Bu_3P in the presence of catalytic amount of $\text{Pd}(\text{Ph}_3\text{P})_4$ to provide a convenient method for synthesis of functionalized cycloalkenes was described by Krische and coworkers [22] (eq. 7).



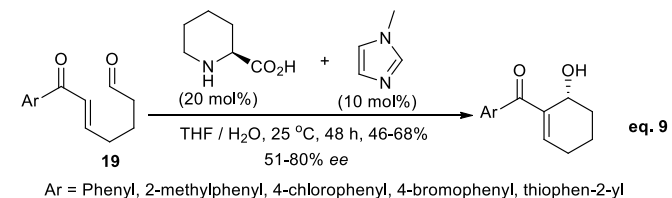
Oshima [23] reported an interesting intramolecular Baylis-Hillman reaction of substrates (17) having both the activated alkene moiety and electrophile component leading to the synthesis of cyclopentene ($\text{R} = \text{SEt}$; $n = 1$) and cyclohexene ($\text{R} = \text{Ph}$; $n = 2$) derivatives (Scheme 9).



For the first time, Hong and coworkers [24] demonstrated an efficient proline catalyzed enantioselective intramolecular Baylis-Hillman reaction of enone-aldehyde system (18) under the influence of imidazole (eq. 8).

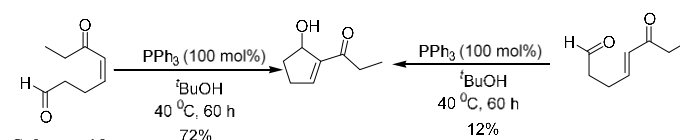


Miller and coworkers [25] reported, the application of (*S*)-2-pipecolic acid for promoting asymmetric intramolecular Baylis-Hillman reaction of enone-aldehyde system (19) in the presence of *N*-methylimidazole to provide the resulting adduct in good enantioselectivities (eq. 9).

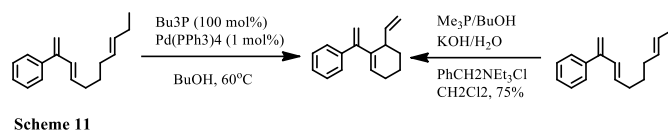


Shi and coworkers [26] reported an interesting intramolecular Baylis-Hillman reaction of enone-aldehyde under the influence of PPh_3 . They have observed the role of stereochemistry of double bond in enone-aldehyde system. Thus, the substrate with (*Z*)- stereochemistry afforded much higher yield than the corresponding substrate with

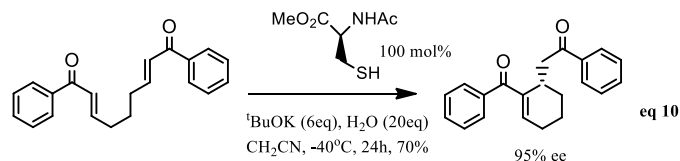
(*E*) stereochemistry (Scheme 10).



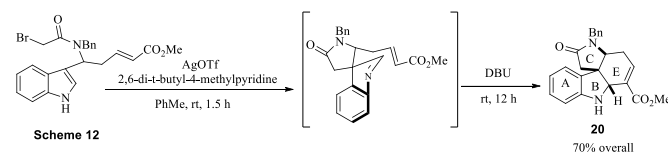
Krische [27] and Krafft [28] have independently reported an interesting intramolecular Baylis-Hillman reaction of the enone-allyl carbonates and enone-allyl alcohol respectively as substrates to provide convenient methodologies for synthesis of functionalized cycloalkenes (Scheme 11).



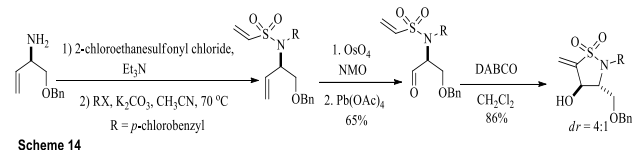
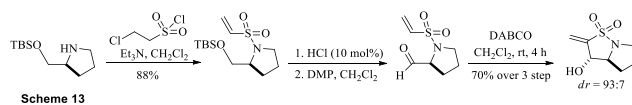
Aroyan and Miller [29] have examined the asymmetric intramolecular Baylis-Hillman reaction of bisenone in the presence of cysteine derivative which provided the corresponding functionalized cyclohexene derivatives in good yields with high enantioselectivities (eq. 10).



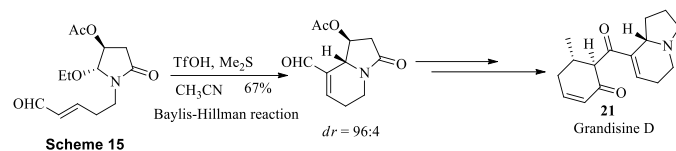
Andrade and Sirasani [30] have published a methodology by a novel sequential one-pot alkylation and intramolecular Baylis-Hillman reaction for synthesis of ABCE tetracyclic framework (20) of *Strychnos* alkaloids (Scheme 12).



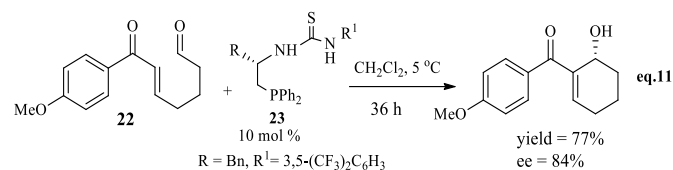
Hanson and coworkers [31, 32] have reported intramolecular Baylis-Hillman reaction as the key step to generate sultams via a strategy involving functional group pairing between vinyl sulfonamide and suitably protected amino alcohol/aldehyde following the reaction sequences (Schemes 13 & 14).



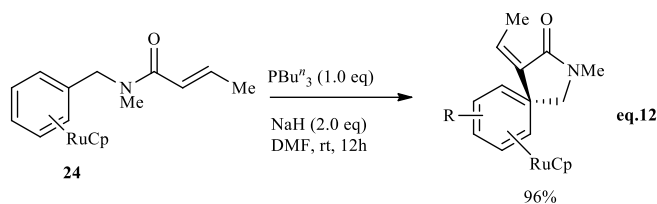
Total synthesis of grandisine D (21) was reported by Tamura and coworkers [33] using Bronsted acid mediated intramolecular Baylis-Hillman reaction as the key step (Scheme 15).



Wu co-workers^[34] demonstrated an interesting intramolecular asymmetric Baylis-Hillman ring closing reaction of enone-aldehyde substrate (**22**) using chiral amino acid derived phosphinothiourea as a catalyst (**23**) (eq. 11).



Pigge and co-workers³⁵ reported an interesting organometallic intramolecular Baylis-Hillman reaction. In this reaction ruthenium-arene complex (**24**) is employed as an electrophile to provide the resulting spiro adduct with 100% diastereoselectivity (eq. 12).



References

- Singh V, Batra S. *Tetrahedron*. 2008; 64:4511.
- Ciganek E. in *Organic Reactions*: (Ed. L. A. Paquette) Wiley: New York. 1997; 51:201.
- Singh V, Batra S. *Tetrahedron*. 2008; 64:4511.
- Baylis AB, Hillman MED. German patent 2155113, *Chem. Abstr.* 1972; 77:34174.
- Hillman MED, Baylis AB. U. S. Patent 3743669, 1973.
- Black GP, Dinon, F, Fratucello S, Murphy PJ, Nielsen M, Williams HL, Walshe NDA. *Tetrahedron Lett.* 1997; 38:8561.
- Dinon F, Richards E, Murphy PJ, Hibbs DE, Hursthouse, MB, Malic KMA. *Tetrahedron Lett.* 1999; 40:3279.
- Richards EL, Murphy PJ, Dinon F, Fratucello S, Brown PM, Gelbrich T, Hursthouse MB. *Tetrahedron*. 2001; 57:7771.
- Wang LC, Luis AL, Agapiou K, Jang, HY, Krische MJ. *J Am. Chem. Soc.* 2002; 124:2402.
- Frank SA, Mergott DJ, Roush WR. *J Am Chem Soc.* 2002; 124:2404.
- Mergott DJ, Frank SA, Roush WR. *Org. Lett.* 2002; 4:3157.
- Keck GE, Welch DS. *Org. Lett.* 2002; 4:3687.
- Yagi K, Turitani T, Shinokubo H, Oshima K. *Org. Lett.* 2002; 4:3111.
- Basavaiah D, Jaganmohan Rao, A. *Chem. Commun.* 2003, 604.
- Methot JL, Roush WR. *Org. Lett.* 2003; 5:4223.
- Keck GE, Welch DS. *Org. Lett.* 2002; 4:3687.
- Yagi K, Turitani T, Shinokubo H, Oshima K. *Org. Lett.* 2002; 4:3111.
- Yeo JE, Yang X, Kim HJ, Koo S. *Chem. Commun.* 2004, 236.
- Krafft ME, Wright JA. *Chem. Commun.* 2006, 2977.
- Krafft ME, Haxell TFN, Seibert KA, Abboud KA. *J Am. Chem. Soc.* 2006; 128:4174.
- Krafft ME, Seibert K.A, Haxell TFN, Hirose C. *Chem. Commun.* 2005, 5772.
- Jellerichs BG, Kong JR, Krische MJ. *J Am. Chem. Soc.* 2003; 125:7758.
- Yagi K, Turitani T, Shinokubo H, Oshima K. *Org. Lett.* 2002; 4:311.
- Chen SH, Hong BC, Su CF, Sarshar S. *Tetrahedron Lett.* 2005; 46:8899.
- Aroyan CE, Vasbinder MM, Miller SJ. *Org. Lett.* 2005; 7: 3849.
- Teng WD, Huang R, Kwong CKW, Shi M, Toy PH. *J Org. Chem.* 2006; 71:368.
- Jellerichs BG, Kong JR, Krische MJ. *J Am. Chem. Soc.* 2003; 125:7758.
- Krafft ME, Haxell TFN. *J Am. Chem. Soc.* 2005, 127; 10168.
- Aroyan CE, Miller SJ. *J Am. Chem. Soc.* 2007; 129:256.
- Zhou A, Hanson PR. *Org. Lett.* 2008; 10:2951.
- Zhou A, Rayabarapu D, Hanson PR. *Org. Lett.* 2009; 11:531.
- Kurasaki H, Okamoto I, Morita N, Tamura O. *Org. Lett.* 2009; 11:1179.
- Adam W, Salgado VON, Peters EM, Peters K. von Schnering, H. G. *Chem. Ber.* 1993; 126:1481.
- Gong JJ, Yuan K, Song HL, Wu XY. *Tetrahedron*. 2010; 66:2439.
- Pigge FC, Dhanya R, Hoefgen ER. *Angew. Chem. Int. Ed.* 2007; 46:2887.